Revision Date 19-May-2015,



CARBON DIOXIDE Safety Data Sheet

1. IDENTIFICATION

Product identifier Product Name

Phone: 317-290-0003 www.indianaoxygen.com CARBON DIOXIDE

Other means of identification Safety data sheet number UN/ID no. Synonyms

IOC-P023 UN1013 LASER Carbon Dioxide, LASER Carbon Dioxide Ultra, Carbonic Anhydride, Carbonic Acid Gas, Carbon Dioxide USP

Recommended use of the chemical and restrictions on useRecommended UseIndustrial and professional use.Uses advised againstConsumer use

Details of the supplier of the safety data sheet Indiana Oxygen 6099 W. Corporate Way Indianapolis, IN 46278

* May include subsidiaries or affiliate companies/divisions.

For additional product information contact your local customer service.

Emergency telephone number Company Phone Number

1-800-535-5053 (Infotrak)

2. HAZARDS IDENTIFICATION

Classification

OSHA Regulatory Status

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200).

Gases under pressure	Liquefied gas
	Yes

Label elements



Signal word

Warning

Hazard Statements Contains gas under pressure; may explode if heated May displace oxygen and cause rapid suffocation

May cause frostbite May increase respiration and heart rate

Precautionary Statements - Prevention Do not handle until all safety precautions have been read and understood Avoid breathing gas Do not get in eyes, on skin, or on clothing Use and store only outdoors or in a well ventilated place Use a backflow preventive device in piping Use only with equipment rated for cylinder pressure Close valve after each use and when empty

Precautionary Statements - Response IF INHALED: Remove person to fresh air and keep comfortable for breathing. Get medical attention/advice. IF ON SKIN:. Get immediate medical advice/attention. Thaw frosted parts with lukewarm water. Do not rub affected area.

Precautionary Statements - Storage Protect from sunlight when ambient temperature exceeds 52°C/125°F

Hazards not otherwise classified (HNOC) Not applicable

3. COMPOSITION/INFORMATION ON INGREDIENTS

Chemical Name	CAS No.	Volume %	Chemical Formula
Carbon dioxide	124-38-9	100	CO 2

4. FIRST AID MEASURES

Description of first aid measures

General advice	Show this safety data sheet to the doctor in attendance.
Inhalation	Remove to fresh air and keep comfortable for breathing. If breathing is difficult, give oxygen. If breathing has stopped, give artificial respiration. Get medical attention immediately.
Skin contact	For dermal contact or suspected frostbite, remove contaminated clothing and flush affected areas with lukewarm water. DO NOT USE HOT WATER. A physican should see the patient promptly if contact with the product has resulted in blistering of the dermal surface or in deep tissue freezing.
Eye contact	If frostbite is suspected, flush eyes with cool water for 15 minutes and obtain immediate medical attention.
Ingestion	Not an expected route of exposure.
Self-protection of the first aider	RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS.
Most important symptoms and effects,	both acute and delayed
Symptoms	Simple asphyxiant. May cause suffocation by displacing the oxygen in the air. Exposure to oxygen-deficient atmosphere (<19.5%) may cause dizziness, drowsiness, nausea, vomiting, excess salivation, diminished mental alertness, loss of consciousness and death. Exposure to atmospheres containing 8-10% or less oxygen will bring about unconsciousness without warning and so quickly that the individuals cannot help or protect themselves. Lack of sufficient oxygen may cause serious injury or death. Depending on concentration and duration of exposure to carbon dioxide may cause increased respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is decreased to 15-17%. Contact with liquid may cause cold burns/frostbite.
Indication of any immediate medical at	tention and special treatment needed
Note to physicians	Treat symptomatically.
	5. FIRE-FIGHTING MEASURES

Suitable extinguishing media

Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

Specific extinguishing methods Continue to cool fire exposed cylinders until flames are extinguished. Damaged cylinders should be handled only by specialists.

<u>Specific hazards arising from the chemical</u> Non-flammable gas. Cylinders may rupture under extreme heat.

Protective equipment and precautions for firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Personal precautions

Evacuate personnel to safe areas. Ensure adequate ventilation, especially in confined areas. Monitor oxygen level. Wear self-contained breathing apparatus when entering area unless

	atmosphere is proved to be safe.
Environmental precautions	
Environmental precautions	Prevent spreading of vapors through sewers, ventilation systems and confined areas.
Methods and material for containment	t and cleaning up
Methods for containment	Stop the flow of gas or remove cylinder to outdoor location if this can be done without risk. If leak is in container or container valve, contact the appropriate emergency telephone number in Section 1 or call your closest Indiana Oxygen Company location.
Methods for cleaning up	Return cylinder to Indiana Oxygen or an authorized distributor.
	7. HANDLING AND STORAGE
Precautions for safe handling	
Advice on safe handling	For applications with moist Carbon Dioxide, 316, 309 and 310 stainless steels may be used as well as Hastelloy® A, B, & C and Monel®. Ferrous nickel alloys are slightly suspectible to corrosion. At normal temperatures carbon dioxide is compatible with most plastics and elastomers.
	Protect cylinders from physical damage; do not drag, roll, slide or drop. When moving cylinders, even for short distance, use a cart designed to transport cylinders. Never attempt to lift a cylinder by its valve protection cap. Never insert an object (e.g. wrench, screwdriver, pry bar,etc.) into valve cap openings. Doing so may damage valve, causing leak to occur. Use an adjustable strap wrench to remove over-tight or rusted caps. Use only with adequate ventilation. Use a backflow preventive device in piping. Use only with equipment rated for cylinder pressure. Close valve after each use and when empty. If user experiences any difficulty operating cylinder valve discontinue use and contact supplier. Ensure the complete gas system has been checked for leaks before use.
	Never put cylinders into trunks of cars or unventilated areas of passenger vehicles. Never attempt to refill a compressed gas cylinder without the owner's written consent. Never strike an arc on a compressed gas cylinder or make a cylinder a part of an electrical circuit.
	Only experienced and properly instructed persons should handle gases under pressure. Always store and handle compressed gas cylinders in accordance with Compressed Gas Association, pamphlet CGA-P1, Safe Handling of Compressed Gases in Containers.
	For additional storage recommendations, consult Compressed Gas Association's Pamphlets P-1,AV-7, G-6, G-6.1, G-6.2, G6.3, G-6.5, G-6.7, G-6.9, PS-5,TB-10, and SB-2.
Conditions for safe storage, including	any incompatibilities
Storage Conditions	Store in cool, dry, well-ventilated area of non-combustible construction away from heavily trafficked areas and emergency exits. Keep at temperatures below 52°C / 125°F. Cylinders should be stored upright with valve protection cap in place and firmly secured to prevent falling. Full and empty cylinders should be segregrated. Use a "first in-first out" inventory system to prevent full cylinders from being stored for excessive periods of time. Stored containers should be periodically checked for general condition and leakage.
Incompatible materials	Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Control parameters

Exposure Guidelines

oosure Guidelines			
Chemical Name	ACGIH TLV	OSHA PEL	NIOSH IDLH
Carbon dioxide	STEL = 30000 ppm	TWA: 5000 ppm	IDLH: 40000 ppm
124-38-9	TWA: 5000 ppm	TWA: 9000 mg/m ³	TWA: 5000 ppm
		(vacated) TWA: 10000 ppm	TWA: 9000 mg/m ³
		(vacated) TWA: 18000 mg/m ³	STEL: 30000 ppm
		(vacated) STEL: 30000 ppm	STEL: 54000 mg/m ³
		(vacated) STEL: 54000 mg/m ³	

Image: Constraint of the second stream of

Other Information	Vacated limits revoked by the Court of Appeals decision in AFL-CIO v. OSHA, 965 F.2d 962 (11th Cir., 1992).
Appropriate engineering controls	
Engineering Controls	Local exhaust ventilation to prevent accumulation of high concentrations and maintain air-oxygen levels at or above 19.5%. Oxygen detectors should be used when asphyxiating gases may be released. Systems under pressure should be regularly checked for leakages.
Individual protection measures, such as	personal protective equipment
Eye/face protection	Wear safety glasses with side shields (or goggles). If splashes are likely to occur, wear:. Goggles. Face-shield.
Skin and body protection	Work gloves and safety shoes are recommended when handling cylinders. Wear cold insulating gloves when handling liquid.
Respiratory protection	Use positive pressure airline respirator with escape cylinder or self contained breathing apparatus for oxygen-deficient atmospheres (<19.5%). If exposure limits are exceeded or irritation is experienced, NIOSH/MSHA approved respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory protection must be provided in accordance with current local regulations.
General Hygiene Considerations	Handle in accordance with good industrial hygiene and safety practice. Do not get in eyes, on skin, or on clothing.

9. PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Physical state Appearance Odor Odor threshold pH Melting point Evaporation rate Lower flammability limit: Upper flammability limit: Flash point Autoignition temperature Decomposition temperature Water solubility	Compressed liquefied gas Colorless. Odorless. No information available No data available -56.6 °C / -69.8 °F Not applicable Not applicable Not applicable Not applicable No data available No data available O.145 g/ml @ 25°C	
Partition coefficient Kinematic viscosity	No data available Not applicable	

Component Level Information:

Chemical Name	Molecular weight	Boiling point	Vapor Pressure	Vapor density (air	Gas Density	Critical
				=1)	kg∕m³@20°C	Temperature

Carbon dioxide	44.01	-78.5 °C	838 psig (5778	1.522	1.839	31.1 °C
		(Sublimes)	kPa) @ 21.1°C			

10. STABILITY AND REACTIVITY

Reactivity

Not reactive under normal conditions

Chemical stability

Stable under normal conditions.

Explosion data

Sensitivity to Mechanical Impact	None.
Sensitivity to Static Discharge	None.

Possibility of Hazardous Reactions None under normal processing.

Conditions to avoid

Due to the presence of Carbon dioxide, Carbonic acid is formed in the presence of moisture.

Incompatible materials

Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

Hazardous Decomposition Products

Oxygen. Carbon monoxide (CO).

11. TOXICOLOGICAL INFORMATION

Information on likely routes of exposure

Inhalation	Acidosis, adrenal cortical exhaustion, and other metabolic stresses have resulted from prolonged continuous exposure to 1-2% carbon dioxide (10,000 ppm-20,000 ppm). The ACGIH TLV of 5,000 ppm is expected to provide a good margin of safety from asphyxiation and undue metabolic stress provided sufficient oxygen levels are maintained in the air. Increased physical activity, duration of exposure, and decreased oxygen content can affect systemic and respiratory effects resulting from exposure to carbon dioxide.
Skin contact	Contact with liquid may cause cold burns/frostbite.
Eye contact	Contact with liquid may cause cold burns/frostbite.
Ingestion	Not an expected route of exposure.
Information on toxicological effects	
Symptoms	Depending on concentration and duration of exposure to carbon dioxide may cause increased respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is decreased to 15-17%.
Delayed and immediate effects as well	as chronic effects from short and long-term exposure
Irritation Sensitization Germ cell mutagenicity Carcinogenicity Reproductive toxicity	Not classified. Not classified. Not classified. This product does not contain any carcinogens or potential carcinogens listed by OSHA, IARC or NTP. Not classified.

STOT - single exposure STOT - repeated exposure Chronic toxicity Target Organ Effects Aspiration hazard Not classified. Not classified. Chronic harmful effects are not known from repeated inhalation of concentrations below PEL/TLV. Central Vascular System (CVS), Respiratory system. Not applicable.

Numerical measures of toxicity

Component Level Information:

Chemical Name	Oral LD50	Dermal LD50	Inhalation LC50	Inhalation LC50 (CGA P-20)
Carbon dioxide 124-38-9	-	-	470,000 ppm (Rat)	-

Product Information Oral LD50 Dermal LD50 Inhalation LC50

No information available No information available TCLo - 10,000 ppm (Rat) 24 hours/30 days-continuous

12. ECOLOGICAL INFORMATION

<u>Ecotoxicity</u> No known acute aquatic toxicity.

Persistence and degradability No information available.

<u>Bioaccumulation</u> No information available.

Global warming potential (GWP)

13. DISPOSAL CONSIDERATIONS

Waste treatment methods

Disposal of wastes

Do not attempt to dispose of residual waste or unused quantities. Return in the shipping container PROPERLY LABELED WITH ANY VALVE OUTLET PLUGS OR CAPS SECURED AND VALVE PROTECTION CAP IN PLACE to Indiana Oxygen Company for proper disposal.

14. TRANSPORT INFORMATION

DOT

UN/ID no.	UN1013
Proper shipping name	Carbon Dioxide
Hazard Class	2.2
Description	UN1013, Carbon dioxide, 2.2
Emergency Response Guide Number	120

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TDG

<u> </u>	
UN/ID no.	UN1013
Proper shipping name	Carbon Dioxide
Hazard Class	2.2
Description	UN1013, Carbon dioxide, 2.2

UN1013

MEX

UN/ID no.

Proper shipping name	Carbon Dioxide
Hazard Class	2.2
Description	UN1013, Carbon dioxide, 2.2

IATA

UN/ID no.	UN1013
Proper shipping name	Carbon Dioxide
Hazard Class	2.2
ERG Code	2L
Description	UN1013, Carbon dioxide, 2.2

IMDG

UN/ID no.	UN1013
Proper shipping name	Carbon dioxide
Hazard Class	2.2
EmS-No.	F-C, S-V
Description	UN1013, Carbon dioxide, 2.2

ADR

UN/ID no.	UN1013
Proper shipping name	Carbon Dioxide
Hazard Class	2.2
Classification code	2A
Tunnel restriction code	(C/E)
Special Provisions	584, 653
Description	UN1013, Carbon dioxide, 2.2, (C/E)

15. REGULATORY INFORMATION

International Inventories	
TSCA	Complies
DSL/NDSL	Complies
EINECS/ELINCS	Complies

Legend:

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List EINECS/ELINCS - European Inventory of Existing Chemical Substances/European List of Notified Chemical Substances

US Federal Regulations

SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372.

SARA 311/312 Hazard Categories	
Acute Health Hazard	No
Chronic Health Hazard	No
Fire Hazard	No
Sudden release of pressure hazard	Yes
Reactive Hazard	No

CERCLA

This material, as supplied, does not contain any substances regulated as hazardous substances under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302) or the Superfund Amendments and Reauthorization Act (SARA) (40 CFR 355). There may be specific reporting requirements at the local, regional, or state level pertaining to releases of this material.

Clean Air Act, Section 112 Hazardous Air Pollutants (HAPs) (see 40 CFR 61)

This product does not contain any substances regulated as hazardous air pollutants (HAPS) under Section 112 of the Clean Air Act Amendments of 1990.

CWA (Clean Water Act)

This product does not contain any substances regulated as pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)

Risk and Process Safety Management Programs

This material, as supplied, does not contain any regulated substances with specified thresholds under 40 CFR Part 68. This product does not contain any substances regulated as Highly Hazardous Chemicals pursuant to the 29 CFR Part 1910.110.

US State Regulations

California Proposition 65

This product does not contain any Proposition 65 chemicals

U.S. State Right-to-Know Regulations

Chemical Name	New Jersey	Massachusetts	Pennsylvania
Carbon dioxide	Х	Х	Х
124-38-9			

International Regulations

Chemical Name	Carcinogenicity	Exposure Limits
Carbon dioxide	-	Mexico: TWA= 5000 ppm
		Mexico: TWA= 9000 mg/m ³
		Mexico: STEL= 15000 ppm
		Mexico: STEL= 27000 mg/m ³

16. OTHER INFORMATION

NFPA	Health hazards 2	Flammability 0	Instability 0	Physical and Chemical
		·	·	Properties Simple
				asphyxiant

Note: Ratings were assigned in accordance with Compressed Gas Association (CGA) guidelines as published in CGA Pamphlet P-19-2009, CGA Recommended Hazard Ratings for Compressed Gases, 3rd Edition.

Issue Date	17-Feb-2015
Revision Date	19-May-2015
Revision Note	Initial Release.

General Disclaimer

For terms and conditions, including limitation of liability, please refer to the purchase agreement in effect between Indiana Oxygen Company (or any of their affiliates and subsidiaries) and the purchaser.

DISCLAIMER OF EXPRESSED AND IMPLIED WARRANTIES

Although reasonable care has been taken in the preparation of this document, we extend no warranties and make no representations as to the accuracy or completeness of the information contained herein, and assume no responsibility regarding the suitability of this information for the user's intended purposes or for the consequences of its use. Each individual should make a determination as to the suitability of the information for their particular purpose(s).

End of Safety Data Sheet